# Energy transfer in complexes of *E. coli* single-stranded DNA-binding protein with single-stranded poly-(2-thiouridylic acid)

Désirée H.H. Tsao, August H. Maki and John W. Chase\*

Department of Chemistry, University of California, Davis, CA 95616 and \*Department of Biochemistry, Case Western Reserve University School of Medicine, Cleveland, OH 44106, USA

### Received 15 December 1989

The complexes of point-mutated Escherichia coli single-stranded DNA-binding protein (Eco SSB) with poly-(2-thiouridylic acid) (poly S<sup>2</sup>U) have been studied by optical detection of magnetic resonance spectroscopy (ODMR). Previous work has determined that two of four tryptophan (Trp) residues in Eco SSB undergo stacking interactions with nucleic acid bases. Selective photoexcitation of S<sup>2</sup>U bases was performed and subsequent triplet—triplet energy transfer from S<sup>2</sup>U to nearby Trp residues in the protein took place. The zero-field splitting (ZFS) parameters and sublevel kinetics were determined for each Trp residue sensitized by S<sup>2</sup>U. The sublevel lifetimes of the two sensitized residues are similar to those of normal Trp. The ZFS parameters, on the other hand, show a dramatic reduction relative to those of the uncomplexed protein, implying a more polarizable environment for the sensitized Trp residues and/or charge transfer interactions with the S<sup>2</sup>U bases.

Spectroscopy, ODMR; DNA binding protein, single-stranded; Energy transfer, triplet-triplet

## 1. INTRODUCTION

Single-stranded DNA binding proteins (SSB) bind to DNA with no sequence specificity and are believed to be involved in many important metabolic functions including repair, replication and recombination [1]. Escherichia coli SSB (Eco SSB) has been the object of extensive study and recently we have employed optical detection of magnetic resonance (ODMR) spectroscopy of the excited triplet state to study complexes of Eco SSB with single-stranded homopolynucleotides [2,3]. From these studies we learned that short range interactions between tryptophan (Trp) residues of the protein and nucleic acid bases play an important role in the stabilization of the complex. Eco SSB has 4 Trp residues (at positions 40, 54, 88 and 135) among its 177 amino acid residues, and based on studies done in complexes of point-mutated Eco SSB, where one Trp residue at a time is substituted by phenylalanine (Phe), it was determined that only Trp40 and Trp54 undergo close range interactions, and probably are stacked with nucleic acid bases [4].

In this study, we report further ODMR investigations of two single point-mutated **Eco** SSB proteins complexed with a unique polynucleotide, poly-(2-thiouridylic acid) (poly S<sup>2</sup>U). The triplet state energy of poly S<sup>2</sup>U is higher than that of Trp. Upon complex formation, we observe triplet-triplet energy transfer from S<sup>2</sup>U bases to Trp, by selectively exciting the triplet state of S<sup>2</sup>U and monitoring Trp emission. By using this method on the

Correspondence address: A.H. Maki, Department of Chemistry, University of California, Davis, CA 95616, USA

mutants Eco SSB W40F and Eco SSB W54F, we have obtained the triplet state zero-field splitting (ZFS) parameters, and the triplet state sublevel kinetics of each sensitized Trp, Trp<sup>54</sup> in the former and Trp<sup>40</sup> in the latter mutated SSB. The low temperature phosphorescence of 2-thiouracil has been studied previously [5], and the triplet state of 1-methyl-2-thiouracil (1-Me-S<sup>2</sup>U) in 1-Me-U has been investigated by ODMR [6]. ODMR results for poly S<sup>2</sup>U are presented in this study.

# 2. MATERIALS AND METHODS

Poly S<sup>2</sup>U was a generous gift from Professor Karl Scheit, Max-Planck-Institut für Biophysikalische Chemie, Göttingen. The point mutated Eco SSB proteins were prepared as described previously [7]. Both protein and poly S<sup>2</sup>U were dialyzed against a 10 mM potassium phosphate buffer, pH 8.4, containing 20 mM NaCl. Poly S2U forms a stable double helix with a melting point of 62.5°C in 50 mM NaCl, pH 7 [8]. At the lower NaCl concentration and higher pH used in this work, the melting point of the polymer  $(3.0 \times 10^{-5} \text{ M})$  in the presence of Eco SSB  $(2.5 \times 10^{-6} \text{ M})$  was lowered to ~44°C while in the absence of the SSB it melted at ~53°C. The complexes for ODMR studies were prepared with a final protein concentration of ~10<sup>-4</sup> M and a 12-fold molar excess of polynucleotide. All samples contained 25% (v/v) of ethylene glycol as cryosolvent. ODMR samples were incubated at 55°C for 30 min and then rapidly frozen by plunging the sample directly into liquid helium. The presence of Eco SSB/poly S<sup>2</sup>U complexes in the chilled sample was verified by the observation of Trp phosphorescence via T --- T energy transfer upon selective excitation of S<sup>2</sup>U.

Phosphorescence and ODMR measurements were conducted using the apparatus described previously [2]. As a control, the pure protein was also studied under the conditions mentioned above using a 16 nm wide excitation band centered at 295 nm. The protein/poly  $S^2U$  complexes, on the other hand, were excited at ~365 nm, selected using a CuSO<sub>4</sub> (12 cm, 50 g/l) filter, a CS-7-60 band pass filter and a WG-345-2 cut off filter. The excitation is in the  $T_1 \leftarrow S_0$  band of

S<sup>2</sup>U ensuring direct excitation of its triplet state. No phosphorescence from Trp was observed when the protein samples were excited under these conditions in the absence of poly S<sup>2</sup>U, since Trp does not absorb significantly in the  $T_1 \leftarrow S_0$  band of  $S^2U$ . Poly  $S^2U$  emits shortlived phosphorescence in the same spectral region as Trp, so time resolved modifications were applied. A shutter routine was used to monitor Trp phosphorescence, in which the sample was excited for 100 ms, followed by a delay of 100 ms prior to the opening of the emission shutter for a period of 300 ms. In order to avoid interference from poly S<sup>2</sup>U phosphorescence, the slow passage ODMR measurements were conducted [9] as follows. After cessation of optical pumping, the sample emission was allowed to decay, during which the microwave frequency was swept. This decay spectrum was then modified by subtracting from it the phosphorescence decay in the absence of a microwave sweep. All modified slow passage responses were corrected for transient effects by extrapolation to zero microwave sweep rate. For the sublevel kinetics measurements, the microwave induced delayed phosphorescence (MIDP) method was employed at 1.2 K, as described previously [2], except that a delay of 350 ms between the closing of excitation and opening of emission shutters was used in order to eliminate all detectable S<sup>2</sup>U emission. The lifetime of the Tx sublevel (the only radiative sublevel of Trp) was obtained by fitting the decay tail of the MIDP response.

### 3. RESULTS AND DISCUSSION

An essential feature of these measurements is to ensure that the Trp residues which we observe are excited exclusively by energy transfer from  $S^2U$ . Excitation of  $S^2U$  in its  $S_1 \leftarrow S_0$  band, which extends to the red of the Trp absorption, could not be done without also exciting Trp. We found, however, that we could produce the Trp triplet state exclusively by energy transfer from  $S^2U$  using direct  $T_1 \leftarrow S_0$  excitation of the latter at  $\sim 365$  nm.

The phosphorescence spectrum of poly S<sup>2</sup>U is shown in fig.1A. The sensitized time-delayed spectrum of the Eco SSB W40F mutant complexed with poly S<sup>2</sup>U is shown in fig.1B. The well resolved vibronic bands are characteristic of Trp phosphorescence. The experimental results obtained for the Trp residues of the SSB mutants and their complexes with poly S<sup>2</sup>U are summarized in table 1. Also included are data for 1-Me-S<sup>2</sup>U [7] and for poly S<sup>2</sup>U. The uncomplexed protein, Eco SSB W88F, W135F, contains only the two Trp residues that are involved in close range interaction with nucleic acid bases upon complex formation. Data on this double mutated SSB and its complex with poly dT [2] are also given in table 1. The phosphorescence 0,0 band of the Eco SSB complexes with poly S<sup>2</sup>U are red-shifted by over 2 nm relative to the uncomplexed protein, but in this case, only the Trp residues that are sensitized by the poly S<sup>2</sup>U bases emit. The lifetimes of the sensitized Trp residues given in table 1 show a slight reduction relative to the uncomplexed protein.

In contrast with the phosphorescence lifetimes which show little change upon interaction of Eco SSB with poly S<sup>2</sup>U, the ZFS of the sensitized Trp residues undergo a pronounced reduction relative to their values in the uncomplexed protein. Both |D| and |E| are greatly reduced, indicating that the average separation of the unpaired electrons increases for the triplet states

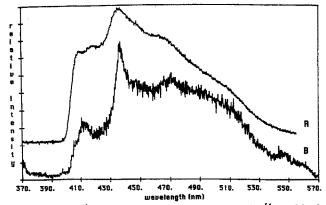


Fig.1. A: Poly  $S^2U$  phosphorescence spectrum. B:  $Trp^{54}$  sensitized phosphorescence in Eco SSB W40F + poly  $S^2U$ . The shutter routine described in the text was used and the sample was excited at  $\sim 365$  nm in the  $S^2U$   $T_1 \longleftarrow S_0$  absorption band. For both A and B, the spectra were obtained at 4.2 K with a scan rate of 20 nm/min.

of both Trp<sup>40</sup> and Trp<sup>54</sup> upon complex formation with poly S<sup>2</sup>U (table 1). Similar effects observed for Trp 54 upon complex formation of **Eco** SSB with poly dT [2,10] have been attributed to the effects of stacking interactions with thymine. The ZFS reductions are considerably larger with poly S<sup>2</sup>U than with poly dT. Stacking interactions with nucleic acid bases, especially those containing polarizable atoms such as S, lead to an increase in the polarizability of the environment and a corresponding expansion of the triplet state wavefunction. This effect leads to a reduction of |D| and |E|. Charge transfer contributions to the triplet state of Trp can also lead to a reduction in the ZFS, particularly in

Table 1

Triplet state properties of point-mutated Eco SSB/poly S<sup>2</sup>U complexes and related systems

| Sample                               | λ <sub>0.0</sub><br>(nm) | Lifetimes <sup>e</sup> (s) |       | (E)   |
|--------------------------------------|--------------------------|----------------------------|-------|-------|
| 1-Me-S <sup>2</sup> U <sup>a</sup>   | 390.0                    | 0.007                      | 8.679 | 2.182 |
|                                      |                          | 0.041                      |       |       |
|                                      |                          | 0.153                      |       |       |
| Poly S <sup>2</sup> U <sup>b</sup>   | 410                      | 0.008                      | 8.26  | 2.18  |
|                                      |                          | 0.026                      |       |       |
|                                      |                          | 0.122                      |       |       |
| Eco SSB W88F, W135F°                 | 410.9                    | 6.1                        | 2.99  | 1.26  |
| Eco SSB W88F, W135F +                |                          |                            |       |       |
| poly dT <sup>d</sup>                 | 415.1                    |                            |       |       |
| Eco SSB W40                          |                          | 5,1                        | 3.03  | 1.29  |
| Eco SSB W54                          |                          | 2.6                        | 2.75  | 1.18  |
| Eco SSB W54F + poly S <sup>2</sup> U | 414                      | 5.5                        | 2.70  | 1.10  |
| Eco SSB W40F + poly S <sup>2</sup> U | 413                      | 5.4                        | 2.64  | 1.11  |

a Polycrystalline sample in 1-Me-U host, except for the lifetimes, which were measured in an ethyl ether/isopentane/ethanol (5:5:2) glass. Data from [7]

b In aqueous buffer, 25% ethylene glycol

Measurements for the protein reported in [2] were repeated in this work in the solvent system reported in the text, because of changes in the microenvironment of Trp

d Data from [2]

<sup>\*</sup> Measurements in this work were made at 4.2 K

the magnitude of |D|. The fact that both |D| and |E| are reduced, and the effect is larger for the more polarizable poly S<sup>2</sup>U than for poly dT, suggests that local polarizability effects are important. Further evidence for the importance of local polarizability in determining |D| and |E| is provided by the complexes of Eco SSB with poly Br<sup>5</sup>U [2] in which the ZFS parameters of Trp<sup>40</sup> and Trp<sup>54</sup> also are reduced relative to their values in the uncomplexed protein.

The triplet state sublevel kinetics of Trp obtained from MIDP measurements in the poly S<sup>2</sup>U complexes are compared with those of Trp [11] in table 2. The sensitized Trps have a relatively long normal lifetime; the sulfur in S<sup>2</sup>U does not produce a noticeable external heavy atom effect (HAE). This is in contrast with complexes of Eco SSB with poly Br5U and poly Hg5U which exhibit large external HAE [2], providing direct evidence for stacking interactions. The magnitude of the HAE, especially with poly Hg<sup>5</sup>U is consistent only with an out of plane location of the heavy atom. The sublevel lifetimes of Trp<sup>40</sup> and Trp<sup>54</sup> in the mutant complexes with S<sup>2</sup>U are very similar, as are the average lifetimes. The phosphorescence decay lifetimes measured at 4.2 K (table 1), when rapid spin-lattice relaxation averages the properties of all 3 sublevels, are in good agreement with the calculated average lifetimes (table 2), obtained from the individual sublevel kinetics measured at 1.2 K.

The triplet states of Trp<sup>40</sup> and Trp<sup>54</sup> in Eco SSB, known to undergo close-range interactions with polynucleotide bases upon complex formation, were populated selectively via triplet-triplet energy transfer in the complex with poly S<sup>2</sup>U by direct excitation of the triplet state of S<sup>2</sup>U. Using Eco SSB mutants W54F and W40F, it was possible to obtain the triplet state ZFS |D| and |E| parameters, as well as the sublevel kinetics for Trp40 and Trp54 individually. The ZFS of these residues were found to be reduced significantly below their values in the uncomplexed protein. Triplet-triplet energy transfer involves the direct transfer of energy from the excited donor to the acceptor via an exchange mechanism [12]. This requires an overlap of their electron clouds. The effective range for this process is ~0.5 nm, showing that a close range interaction prob-

Table 2

Sublevel triplet state kinetics of Trp<sup>40</sup> and Trp<sup>54</sup> sensitized by poly S<sup>2</sup>U complexed with single point-mutated Eco SSB proteins

| Sample                               | k <sub>x</sub> |       | $k_z$ (s <sup>-1</sup> ) | k <sub>av</sub> a | τ <sub>av</sub><br>(s) |
|--------------------------------------|----------------|-------|--------------------------|-------------------|------------------------|
| Eco SSB W54F + poly S <sup>2</sup> U | 0.322          | 0.112 | 0.077                    | 0.170             | 5.9                    |
| Eco SSB W40F + poly S <sup>2</sup> U | 0.333          | 0.116 | 0.093                    | 0.180             | 5.6                    |
| Trp <sup>b</sup>                     | 0.24           | 0.119 | 0.038                    | 0.134             | 7.5                    |

 $a_{av} = \frac{(k_x + k_y + k_z)}{3} = \frac{1}{\tau_{av}}$ 

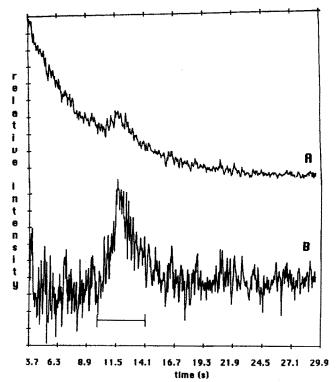


Fig. 2. Delayed slow passage of |D|-|E| ODMR transition of Trp<sup>40</sup> in Eco SSBW54F + poly S<sup>2</sup>U. A: Microwaves are swept from 1.50 to 2.00 GHz (shown by horizontal bar) as the sample phosphorescence decays. B: Same ODMR transition after subtraction of the free decay. The sweep rate was 125 MHz/s, averaged over 22 scans.

ably involving stacking exists between the Trp<sup>40</sup> and Trp<sup>54</sup> residues and polynucleotide bases.

Acknowledgements: We thank Professor Dr Karl Heinz Scheit for providing us with a generous sample of poly S<sup>2</sup>U and Bruce Schlyer for his valuable help in some of the experiments. This work was supported by the NIH grants ES-02662 to A.H.M. and GM-11301 and CA-13330 to J.W.C.

## REFERENCES

- Chase, J.W. and Williams, K.R. (1986) Annu. Rev. Biochem. 55, 103-136.
- [2] Tsao, D.H.H., Casas-Finet, J.R., Maki, A.H. and Chase, J.W. (1989) Biophys. J. 55, 927-936.
- [3] Khamis, M.I., Casas-Finet, J.R., Maki, A.H., Murphy, J.B. and Chase, J.W. (1987) FEBS Lett. 220, 347-352.
- [4] Khamis, M.I., Casas-Finet, J.R., Maki, A.H., Murphy, J.B. and Chase, J.W. (1987) J. Biol. Chem. 262, 10938-10945.
- [5] Pownall, H.J., Schaffer, A.M., Becker, R.S. and Mantulin, W.M. (1978) Photochem. Photobiol. 27, 625-628.
- [6] Taherian, M.R. and Maki, A.H. (1981) Chem. Phys. 55, 85-96.
- [7] Mark, D.F., Lu, S.D., Creasey, A.A., Yamamoto, R. and Lin, L.S. (1984) Proc. Natl. Acad. Sci. USA 81, 5662-5666.
- [8] Bahr, W., Faerber, P. and Scheit, K.H. (1973) Eur. J. Biochem. 33, 535-544.
- [9] Schlyer, B.D., Maki, A.H. and Visser, A.J.W.G. (1989) Chem. Phys. Lett. 154, 39-47.
- [10] Zang, L.-H., Maki, A.H., Murphy, J.B. and Chase, J.W. (1987) Biophys. J. 52, 867-872.
- [11] Zuclich, J., Von Schütz, J.U. and Maki, A.H. (1974) Mol. Phys. 28, 33-47.
- [12] Dexter, D.L. (1953) J. Chem. Phys. 21, 836-850.

b In ethylene glycol/water (1:1, v/v), pH = 7. From [11]